General Chemistry

Protons, Neutrons & Electrons In Class Practice

1.	How many protons does silver (Ag) have?
2.	How many electrons does a neutral atom of rubidium (Rb) have?
3.	An atom of carbon has 8 neutrons; what is its mass number?

- 4. How many neutrons does mercury have if its mass number is 201?______
- 5. Fill in the table below:

Isotope Notation	Protons	Electrons	Neutrons
N-14			
N-15			
U-234			
U-235			
U-238			

6. Fill in the missing parts of the table below:

Name	Symbol	Atomic Number	Mass Number	Protons	Electrons	Neutrons	Isotope Notation
Tin						68	
		39	89				
	Ca		40				
					58	82	
							Ti-48

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